Ecological Assessment of Organochlorine Pesticides in the Surface Water from Huaihe River, China

Chunnian Da, Mei Yue, and Shengping Liu

Abstract—Thirty-four surface water samples were collected from Huaihe River, China. The organic compounds were extracted by solid-phase extraction (SPE) system and detected by gas chromatography (GC) using a mass-selective detector. HCHs were the dominant organochlorine pesticides in surface water. The results of PCA suggested there was some similar interrelation between individual OCPs. Health risks assessment indicated that the risks of children were higher than adults both carcinogenic risk and non-carcinogenic risk. The carcinogenic risk analysis indicated the health risks were very low for children, and there were not carcinogenic risks for adult.

 $\it Index\ Terms{\rm --HCH},\ DDT,\ health\ risk\ assessment,\ Huaihe\ River.$

I. INTRODUCTION

Organochlorine pesticides (OCPs), which are typical POPs, have caused worldwide concern due to their high toxicity, persistence, semi-volatility, and bioconcentration [1], [2]. They can get into the aquatic environment through effluent discharge, agricultural runoff and atmospheric deposition. Information on OCPs concentrations in aquatic water and the importance of evaluating human exposure have been reported in many lakes and river, such as, Poyang lake, Chaohu Lake, Honghu Lake and Yangtze River [3], [4]. To our knowledge, there are few studies on the assessment of OCPs in water from the Huaihe River. Huaihe River, one of the seven largest rivers in China, flows through five provinces of Henan, Hubei, Anhui, Jiangsu and Shandong. The river has become more and more vulnerable to water pollution due to increasing discharges of industrial and agriculture waste along with rapid development of the local economy [5]. This study aims to carry out a survey of OCPs in water from the Huaihe River, China. The levels of pollutants have been determined, and the environmental quality and the impact of prior usage of OCPs in the agricultural and industrial would be assessed.

II. MATERIALS AND METHODS

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A. Sampling

A total of thirty-four surface water samples were collected in July 2015. Fig. 1 showed the map of sampling locations. 5 L water was collected from each sampling site, and transported to the laboratory and kept at -20 $^{\circ}$ C in the refrigerator until further study.

B. Sample Preparation and Extraction

1L water sample was leached via glass fiber, using a vacuum pump. 4, 4'-dichlorobiphenyl was added to the water as a recovery indicator. The organic compounds in each water sample were extracted by solid-phase extraction (SPE) system. Before the extraction, the SPE cartridges were first washed with 5 ml dichloromethane, 5 ml methanol and 5 ml ultrapure water. The cartridges could not be dried when the sample was loaded. About 5 g anhydrous sodium sulfate was connected every cartridge. The water sample was extracted with 10 ml dichloromethane solution for each elution, and every sample was eluted three times. The extracts were concentrated to 1 ml with a rotary evaporator. 10 mL hexane was added as a change solvent, and the extracts were again concentrated to 1 mL and transferred to vial until analysis.

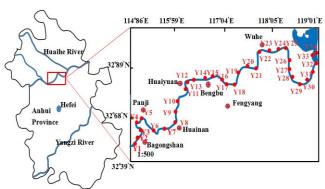


Fig. 1. Locations of sampling points.

C. Instrumental Analysis

341

Sample analysis was conducted on an Agilent 6890 Series GC System connected to an Agilent 5973 Network Mass Selective Detector. A DB-5 MS fused silica capillary column (30m×0.25mm×0.25mm) was used for separation. Helium gas (99.999%) was used as the carrier gas. The samples (1 μ L) were injected automatically by splitless mode. The MS was operated in EI $^+$ mode with selected ion. Electron energy was 70 eV. The column flow was maintained at 1 mL min $^{-1}$. The temperature change was programmed as follows: began at 80 °C for 1 min, ramping at a rate of 12 °C min $^{-1}$ to 200 °C for 10 min, then to 220 °C at 1 °C min $^{-1}$ and kept for 5 min and finally to 290 °C at 15 °C min $^{-1}$ withholding for 5 min.

III. RESULTS AND DISCUSSION

A. Concentration Levels of Organochlorine Pesticides in Surface Water

Concentrations of HCHs and DDTs in surface water are shown in Table I. The detection frequencies for the HCHs and DDTs are 87.5% and 21.56%, respectively, indicating a wide event of HCHs in surface water from the Huaihe River. The average values of \(\text{EHCHs} \) and DDTs are 13.71 ng/L and 0.077 ng/L, respectively. The concentrations of HCHs were much higher than those of the DDTs. Overall; HCHs was the dominant organochlorine pesticides in water samples. Compared with OCPs in the sediments, in our previous study [6], the detection frequencies and the average values of the HCHs increased, however, the detection frequencies and the average values of the DDTs decreased. This is explained by that fact that DDTs have lower water solubility, lower vapour pressure and higher lipophilicity [7]; thus, DDTs were easier to remain in the particulate phase than HCHs [8]. In addition,

the amount of HCHs used to be significantly larger than that of DDTs in the history of China [9]. Also, HCHs were banned a little later than DDTs in this study area.

B. Inter-relation between Individual OCPs

PCA is used to explore the relationship of the measured parameters and contribution of every contaminant among all the contaminants [10]. The results of PCA were presented in Fig 2. The three principal components explained 59.02% of the total variance. The high positive loadings of $\alpha\text{-HCH}$, $\beta\text{-HCH}$, $\gamma\text{-}$ HCH and o'p-DDE in the same group of PC1 (22.26%) recommend the same origin of these pesticides. PC2 accounting for 21.80% of the total variability was mainly related with p'p-DDE, o'p-DDT, o'p-DDD. These pesticides residues in the same group of PC2 could suggest the event status of them in the coastal areas. PC2 reflected the presence of these historical inputs of DDT and its degradation products in this area. As for PC3 (14.97%), the loadings of $\delta\text{-HCH}$ and p'p-DDD suggest these pesticides could have similar migration characteristics.

Sample code	α-НСН	β-НСН	ү-НСН	δ-НСН	∑НСН	o'p-DDE	p'p-DDE	o'p-DDT	p'p-DDT	p'p-DDD	o'p-DDD	∑DDT
S1	2.12	5.07	1.05	0.03	8.27	n.d.	0.04	n.d.	n.d.	n.d.	n.d.	0.04
S2	5.73	24.39	11.32	n.d.	41.44	0.12	n.d.	n.d.	0.013	n.d.	n.d.	0.133
S 3	5.18	13.02	4.13	n.d.	22.33	n.d.	n.d.	n.d.	n.d.	0.19	n.d.	0.19
S4	2.12	11.28	3.21	0.12	16.73	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	0
S5	1.79	15.26	2.85	0.06	19.96	n.d.	n.d.	n.d.	0.163	n.d.	n.d.	0.163
S6	1.02	4.72	1.76	n.d.	7.5	n.d.	0.011	0.11	n.d.	n.d.	n.d.	0.121
S 7	1.11	5.54	2.36	n.d.	9.01	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	0
S8	1.43	11.20	2.16	0.22	15.01	n.d.	n.d.	n.d	n.d	n.d	n.d.	0
S9	1.34	5.43	1.85	0.08	8.7	n.d.	n.d.	n.d.	n.d	n.d.	0.011	0.011
S10	1.91	13.9	2.87	0.09	18.77	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	0
S11	n.d.	12.99	2.67	n.d.	15.66	0.012	n.d.	n.d	n.d.	0.013	n.d.	0.013
S12	1.34	5.5	1.81	0.16	8.81	0.041	n.d.	n.d.	n.d.	n.d.	n.d.	0
S13	1.67	5.64	2.74	0.11	10.16	n.d.	n.d.	n.d.	n.d.	0.013	n.d.	0.013
S14	1.88	3.91	2.36	0.20	8.35	n.d.	0.013	n.d.	n.d.	n.d.	n.d.	0.013
S15	1.45	3.53	2.56	n.d.	7.54	n.d.	0.03	n.d.	0.12	n.d.	0.02	0.17
S16	0.94	3.09	1.87	n.d.	5.9	n.d.	0.25	0.11	n.d.	n.d.	0.12	0.48
S17	1.22	5.07	2.15	0.13	8.57	0.13	n.d.	0.04	0.09	n.d.	0.05	0.18
S18	n.d.	4.39	1.32	n.d.	5.71	n.d.	n.d.	n.d.	0.06	n.d.	n.d.	0.06
S19	0.28	3.02	1.13	n.d.	4.43	0.014	n.d.	0.05	n.d.	n.d.	n.d.	0.05
S20	1.12	11.28	2.11	0.12	14.63	n.d.	n.d.	n.d.	n.d.	n.d.	0.012	0.012
S21	1.19	10.16	1.85	0.05	13.25	n.d.	n.d.	0.03	0.013	n.d.	n.d.	0.043
S22	2.43	4.12	2.79	n.d.	9.34	n.d.	0.013	0.15	n.d.	n.d.	n.d.	0.163
S23	3.18	3.59	4.36	n.d.	11.13	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	0
S24	3.43	10.45	3.19	0.13	17.2	n.d.	n.d.	0.012	n.d.	0.31	n.d.	0.322
S25	5.54	6.43	1.95	0.01	13.93	n.d.	n.d.	n.d.	0.02	n.d.	0.01	0.03
S26	2.91	13.9	2.57	0.03	19.41	n.d.	n.d.	n.d.	n.d.	0.017	n.d.	0.017
S27	2.11	11.98	2.67	n.d.	16.76	n.d.	n.d.	0.01	n.d.	0.013	n.d.	0.023
S28	2.34	5.56	1.91	0.06	9.87	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	0
S29	0.67	4.64	2.07	0.13	7.51	n.d.	n.d.	n.d.	n.d.	0.019	n.d.	0.019

S30	2.88	13.91	2.96	0.26	20.01	n.d.	0.008	n.d.	n.d.	n.d.	0.001	0.009
S31	3.45	3.53	4.56	n.d.	11.54	0.06	n.d.	n.d.	n.d.	n.d.	0.02	0.02
S32	1.94	2.89	2.47	n.d.	7.3	n.d.	n.d.	0.11	n.d.	n.d.	n.d.	0.11
S33	1.02	15.07	8.05	0.92	25.06	n.d.	0.01	n.d.	n.d.	0.13	0.05	0.19
S34	4.73	20.39	1.32	n.d.	26.44	n.d.	n.d.	n.d.	0.01	n.d.	n.d.	0.01

C. Health Risks Assessment

Health risks assessment model

Exposure routes of OCPs are the food chain, drinking water and steam absorption. This study uses a method which was proposed by USEPA [11] to calculate the exposure dose. Briefly, the exposed population is divided into two groups of children (0-6 years old) and adult (> 18 years old), the calculation formulas are as follows:

$$CDI = (C \times IR \times EF \times ED) / (BW \times AT)$$
 (1)

$$R=CDI \times SF$$
 (2)

$$HI = CDI / RfD$$
 (3)

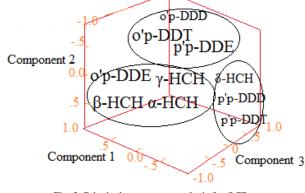
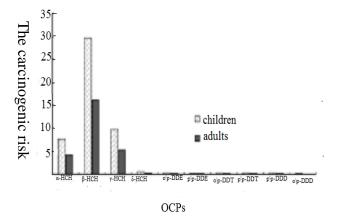


Fig. 2. Principal component analysis for OCPs.



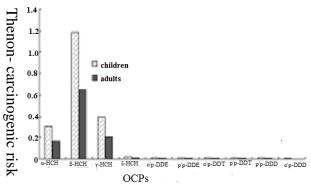


Fig. 3. Thearcinogenic/non-carcinogenic risk (×10⁻⁶) for OCPs.

CDI is exposure dose of long time; C is the concentration of OCPs (mg L⁻¹); IR is daily drinking dose (children: 1 L d⁻¹, adult: 2.2 L d⁻¹);EF is the exposure frequency (350 d a⁻¹);ED is the exposure duration (children: 6 a, adult: 76.27 a);BW is the average body weight (children: 14 kg, adult: 60 kg);AT is the averaging time (children: 2190 d, adult: 26280 d);R is the carcinogenic risk; SF is cancer slope factor (2 kg d mg⁻¹);HI is the noncarcinogenic risk; RfD is Reference Dose (0.02 mg kg⁻¹ d⁻¹);

Health risks assessment

According the above equation (2), (3), (4), we calculated annual averaged carcinogenic/noncarcinogenic risk caused by OCPs in surface water from the Huaihe River. The risk of the different compounds was showed in Table 4. The carcinogenic risk and non-carcinogenic risk of adults and children were showed in Fig. 3. As seen from the Fig .3, the risks of children were higher than adults both carcinogenic risk and non-carcinogenic risk. The results were consistent with the previous study [12, [13].

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